

Green Solvents Screening for the Fractionation of LignoBoost Lignin

Omar Abdelaziz^{1,*}, Jens Prothmann², Charlotta Turner², Christian Hulteberg¹

¹Department of Chemical Engineering, Lund University, P.O. Box 124, SE-221 00 Lund, Sweden ²Department of Chemistry, Centre for Analysis and Synthesis, Lund University, P.O. Box 124, SE-221 00 Lund, Sweden

Abstract

A new two-step solvent-based lignin fractionation method is proposed in the present contribution. Green solvents are screened for their ability to solubilise and extract LignoBoost lignin and the employed solvents mixtures are selected based on the Hansen solubility parameters (HSPs) theory. Selective extraction of LignoBoost lignin is performed at ambient temperature exploiting solvents mixtures of ethvl ethyl (90:10), acetate/methanol (60:40), acetate/methanol acetone/water (70:30), and acetone/water (30:70). Fractionated lignin samples are characterised using various analytical techniques such as Brunauer–Emmett–Teller N_2 adsorption, gel permeation chromatography, Fourier transform infrared spectroscopy, nuclear magnetic resonance spectroscopy, and liquid chromatography/mass spectrometry. The elucidation of such technical lignin physicochemical properties through a sustainable fractionation approach is crucial towards better understanding its complex molecular structure and fostering technologies for advanced lignin-based materials.

Introduction

Lignin is one of the primary constituents of terrestrial plant biomass, together with the carbohydrate polymers cellulose and hemicellulose. It is considered to be the second most abundant biopolymer available on Earth, representing a promising source of renewable novel materials and value-added products [1]. Despite its considerable potential as a raw material for the chemical industries and future biorefineries, lignin remains the most inadequately utilised in comparison to other lignocellulosic biopolymers [2]. Also, its molecular structure is still not well-defined, creating many challenges for its efficient valorisation [3]. In light of this, lignin refining via fractionation appears to be an effective approach towards producing more homogeneous starting raw materials [4].

Materials and Methods

LignoBoost lignin was obtained from Innventia's pilot plant (Bäckhammar, Sweden) and all other chemicals were purchased from Sigma-Aldrich. Table 1 reports the HSPs for solvents mixtures designed for the fractionation experiments. The fractionated/unfractionated lignin samples were characterised using BET, GPC, FT-IR, ¹H-NMR, and LC-MS.

Table 1. Hansen solubility parameters for the solvents mixtures used in the study (where δ_D is the dispersive, δ_P is the polar and δ_H is the hydrogen bonding interactions of HSPs).

Solvent mixture	δ _D (MPa ^{1/2})	δ _P (MPa ^{1/2})	δ _H (MPa ^{1/2})		
EtAc/MeOH 90:10	15.7	6.0	8.7		
EtAc/MeOH 60:40	15.5	8.1	13.2		
Acetone/Water 70:30	15.5	12.1	17.6		
Acetone/Water 30:70	15.5	14.3	31.7		

Conclusions

A new method for technical lignin selective fractionation has been demonstrated using selected green solvents mixtures with comprehensive characterisation of obtained fractions. The developed approach can readily generate more usable lignin fractions, facilitating its further manipulation in different high-value applications.

References

1. Sjöström E (1993) Wood Chemistry: Fundamentals and Applications, 2nd ed. Academic Press, Inc., San Diego, California

2. Ragauskas AJ, Beckham GT, Biddy MJ, et al (2014) Lignin valorization: improving lignin processing in the biorefinery. Science 344:1246843

3. Abdelaziz OY, Hulteberg CP (2016) Physicochemical Characterisation of Technical Lignins for Their Potential Valorisation. Waste and Biomass Valorization. doi: 10.1007/s12649-016-9643-9

4. Boeriu CG, Fiţigău FI, Gosselink RJA, Frissen AE, Stoutjesdijk J, Peter F (2014) Fractionation of five technical lignins by selective extraction in green solvents and characterisation of isolated fractions. Ind Crops Prod 62:481–490

*Omar.Abdelaziz@chemeng.lth.se; Tel.: +46 46 222 9457; fax: +46 46 222 4526 (http://www.lignin.lu.se/)

Results and Discussion

Table 2 summarises the results generated from the N₂ adsorptiondesorption experiments, while Figure 1 shows the gel permeation chromatography curves for the unfractionated/fractionated lignin samples. Figure 2 also presents the FT-IR spectra for the tested samples. The isolation procedure and solvent choice affected the lignin's properties and structure, suggesting notable variations in different obtained lignin fractions.

Table	2.	Surface	area	and	pore	structure	parameters	of	fractionated	
Lignol	Table 2. Surface area and pore structure parameters of fractionated LignoBoost lignin samples determined by BET analysis. Image: Surface									

Sample	Surface area (m ² /g)				Pore volume (cm ³ /g × 10 ³)			Pore size (Å)	
	SBET	SLSA	S _{BJH}	Smicro	V _{total}	V _{BJH}	V _{micro}	D _{BET}	D _{BJH}
Unfractionated	0.40	0.54	0.31	0.26	1.96	1.87	0.11	199.0	240.0
EtAc/MeOH 90:10	3.16	6.96	1.70	0.33	2.33	1.99	0.63	29.53	46.77
EtAc/MeOH 60:40	1.53	2.29	0.03	1.62	0.54	0.54	0.67	14.18	762.98
Acetone/Water 70:30	1.84	2.65	1.98	0.06	7.72	7.44	0.05	167.36	150.62
Acetone/Water 30:70	1.73	2.39	1.89	0.58	6.19	5.98	0.25	142.87	126.57

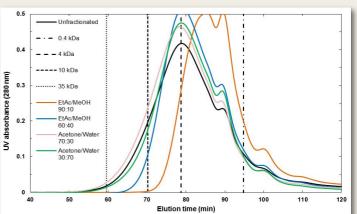


Figure 1. Gel permeation chromatograms for fractionated lignin samples with characteristic UV absorption at 280 nm. The vertical lines indicate the molecular mass (kDa) of the PEG standards that were used for calibration.

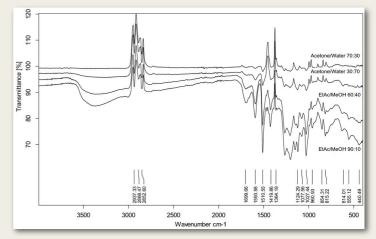


Figure 2. FT-IR spectra of fractionated lignin samples with the wavenumber range 4,000-400 $\mbox{cm}^{-1}.$

Acknowledgements

This research is financed by the Swedish Foundation for Strategic Research (SSF), Grant RBP14-0052

